

# Material Safety Data Sheet **Hydrated Lime**

#### **SECTION 1: PRODUCT AND COMPANY IDENTIFICATION**

**Product Name: Hvdrated Lime** 

Synonym/s: Hydrate; High Calcium Hydrated Lime; HL

Chemical Name: Calcium hydroxide Chemical Formula: Ca(OH)<sub>2</sub>

Product Use/s: Water treatment, pH adjustment, FGT, Construction, Pulp/Paper

## **SECTION 2: HAZARDS IDENTIFICATION**

Hydrated lime is an odorless white powder. Contact can cause irritation to eyes, skin, **Emergency Overview:** 

respiratory system, and gastrointestinal tract.

**Hazard Pictograms:** 





Potential Health Effects

Contact can cause severe irritation or burning of eyes, including permanent damage. Eyes:

Skin: Contact can cause severe irritation or burning of skin, especially in the presence of moisture.

Ingestion: This product can cause severe irritation or burning of gastrointestinal tract if swallowed.

Inhalation: This product can cause severe irritation of the respiratory system. Long-term exposure may

cause permanent damage. Hydrated lime is not listed by MSHA, OSHA, or IARC as a carcinogen. However, this product may contain trace amounts of crystalline silica in the form of quartz or crystobalite, which has been classified by IARC as a Group I carcinogen to humans when inhaled. Inhalation of silica can also cause a chronic lung disorder, silicosis.

Potential Environmental This material is alkaline and if released into water or moist soil will cause an increase in pH.

Effects:

#### SECTION 3: COMPOSITION / INFORMATION ON INGREDIENTS

Ingredient	Chemical Formula	Common Name	Conc. (%)	CAS
Calcium Hydroxide	Ca(OH) <sub>2</sub>	Hydrated Lime	> 90	1305-62-0
Magnesium Oxide	MgO	Periclase	< 3	1309-48-4
Calcium Carbonate	CaCO <sub>3</sub>	Limestone	< 3	1317-65-3
Crystalline Silica	SiO <sub>2</sub>	Quartz	< 2	14808-60-7

(Crystalline Silica is reported as total silica and not just the respirable fraction)

# **SECTION 4: FIRST AID MEASURES**

Immediately flush eyes with generous amounts of water or eye wash solution if water is Eyes:

> unavailable. Pull back eyelid while flushing to ensure that all hydrated lime dust has been washed out. Seek medical attention promptly if the initial flushing of the eyes does not

remove the irritant. Do not rub eyes.

Skin: Brush off or remove as much dry lime as possible. Wash exposed area with large amounts

of water. If burned seriously or if irritation persists, seek medical attention promptly.

Inhalation: Move victim to fresh air. Seek medical attention. If breathing has stopped, give artificial

respiration.

Ingestion: Do not induce vomiting. Seek medical attention immediately. Never give anything by mouth

unless instructed to do so by medical personnel.

Contact may aggravate disorders of the eyes, skin, gastrointestinal tract, and respiratory

**Medical Conditions** 

system.

Aggravated by Exposure:

#### **SECTION 5: FIREFIGHTING MEASURES**

Fire Hazards: Hydrated lime is not combustible or flammable. However, it reacts vigorously with acids, and

> may release heat sufficient to ignite combustible materials in specific instances. Hydrated lime is not considered to be an explosion hazard, although reaction with acids or other

incompatible materials may rupture containers.

Suitable Extinguishing

Media:

Use dry chemical or CO<sub>2</sub> fire extinguisher to extinguish the surrounding fire.

Fire Keep personnel away from and upwind of fire. Avoid skin contact or inhalation of dust. Wear

Fighting Instructions: full fire-fighting turn-out gear (full Bunker gear), and respiratory protection (SCBA).

Hazardous

Not applicable

**Combustion Products:** 

## **SECTION 6: ACCIDENTAL RELEASE MEASURES**

Spill / Leak Procedures: Do Not use water on bulk material spills.

Use proper personal protective equipment.

Small Spills: Use dry methods to collect spilled materials. Avoid generating dust. Do not clean up with

compressed air. Store collected materials in dry, sealed plastic or non-aluminum metal

containers. Residue on surfaces may be water washed.

Large Spills: Use dry methods to collect spilled materials. Evacuate area downwind of clean-up

operations to minimize dust exposure. Store spilled materials in dry, sealed plastic or

non-aluminum metal containers.

Containment: Minimize dust generation and prevent bulk release to sewers or waterways.

Residual amounts of material can be flushed with large amounts of water. Equipment can Clean-up:

be washed with either a mild vinegar and water solution, or detergent and water.

# **SECTION 7: HANDLING AND STORAGE**

Handling: Keep in tightly closed plastic or non-aluminum metal containers. Protect containers from

physical damage. Avoid direct skin contact with the material. Avoid breathing any dust.

Storage: Store in a cool, dry, and well-ventilated location. Do not store near acids or other

incompatible materials. Keep away from moisture. Do not store or ship in aluminum

containers.

## **SECTION 8: EXPOSURE CONTROLS / PERSONAL PROTECTION**

Ingredient	OSHA PEL, TWA 8/40h (mg/m3)	ACGIH TLV, TWA 8/40h (mg/m3)	NIOSH REL, TWA 8/40h (mg/m3)	NIOSH IDLH (mg/m3)
Calcium Hydroxide	15 (total dust) 5 (respirable)	5	5	n/a
Magnesium Oxide	10	10	n/a	n/a
Calcium Carbonate	15 (total dust) 5 (respirable)	10	10 (total dust) 5 (respirable)	n/a
Crystalline Silica	10/(SiO2% + 2) (respirable)	0.025 (respirable)	0.05 (respirable)	50

Engineering Controls: Provide ventilation adequate to maintain PELs.

Respiratory Protection: Use NIOSH/MSHA approved respirators if airborne concentration exceeds PELs.

**Skin Protection:** Use appropriate gloves and footwear to prevent skin contact and the potential for burns.

Clothing should fully cover arms and legs. Should lime get inside clothing or gloves, remove

the clothing and the lime promptly.

Eye Protection: Use safety glasses with side shields or safety goggles. Contact lenses should not be worn

when working with lime products.

Other: Eye wash fountain/stations and emergency showers should be available.

#### **SECTION 9: PHYSICAL AND CHEMICAL PROPERTIES**

Appearance: White free flo	wing powder	Odor: Odorless		Physical State:	Solid
Melting Point (°C/°F):	dec 580/ 1076	Boiling Point (	°C/°F): n/a	Bulk Density: 25	5-32 lbs/ ft3
Specific Gravity	(Apparent) g/cc: 0.4 - 0.55 (True) g/cc: 2.2 - 2.4				
Vapor Pressure (mm Hg): n/a		Vapor Density: n/a		Evaporation Rate: n/a	
pH (25°C/77°F): 12.4	Solubility in Water: Slightly soluble in water at 1.02 g/L at 25 °C				

#### **SECTION 10: STABILITY AND REACTIVITY**

Stability: Chemically stable, but slowly reacts with CO<sub>2</sub> to form calcium carbonate.

See also Incompatibility below.

Hazardous Decomposition/

Products: Does not occur Polymerization: Does not occur

Incompatibility/ Hydrated lime should not be mixed or stored with the following materials, due to the potential for vigorous reaction and release of heat:

Acids (unless in a controlled process)

Reactive Fluoridated Compounds

Reactive Brominated Compounds

Reactive Powdered Metals

Organic Acid Anhydrides

Nitro-Organic Compounds

Reactive Phosphorous Compounds

Interhalogenated Compounds

# SECTION 11: TOXICOLOGICAL INFORMATION

ORL-RAT LD50: 7,340 MG/KG ORL-MUS LD50: 7,300 MG/KG

Hazardous

Hydrated lime is not listed by MSHA, OSHA, or IARC as a carcinogen, but this product may contain trace amounts of crystalline silica, which has been classified by IARC as carcinogenic to humans when inhaled in the form of quartz or crystobalite.

Inhalation, skin and eye contact are the most likely routes of exposure. This material is irritating to the skin and severely irritating to the eyes.

#### **SECTION 12: ECOLOGICAL INFORMATION**

Ecotoxicity: Because of the high pH of this product, it would be expected to produce significant

ecotoxicity upon exposure to aquatic organisms and aquatic systems in high

concentrations (> 1 g/L).

**Environmental Fate:** This material shows no bioaccumulation effect or food chain concentration toxicity.

High pH values will rapidly decrease over time as a result of recarbonation.

This material may be used in soil stabilization or remediation and will show very little

mobility in soils.

#### SECTION 13: DISPOSAL CONSIDERATIONS

Dispose of in accordance with all applicable federal, state, and local environmental regulations. If this product as supplied, and unmixed, becomes a waste, it will not meet the criteria of a hazardous waste as defined under the U.S. Resource Conservation and Recovery Act (RCRA).

#### **SECTION 14: TRANSPORTATION INFORMATION**

Hydrated lime is not classified as a hazardous material by US DOT and is not regulated by the Transportation of Dangerous Goods (TDG) when shipped by any mode of transport.

# **SECTION 15: REGULATORY INFORMATION**

U.S. EPA Regulations: RCRA Hazardous Waste Number (40 CFR 261.33): not listed

RCRA Hazardous Waste Classification (40 CFR 261): not classified

CERCLA Hazardous Substance (40 CFR 302.4) unlisted specific per RCRA, Sec. 3001;

CWA, Sec. 311(b)(4); CWA, Sec. 307(a), CAA, Sec. 112

CERCLA Reportable Quantity (RQ), not listed

SARA 311/312 Codes: not listed

SARA Toxic Chemical (40 CFR 372.65): not listed

SARA EHS (Extremely Hazardous Substance) (40 CFR 355): not listed, Threshold

Planning Quantity (TPQ): not listed

All chemical ingredients are listed on the US EPA TSCA Inventory List.

OSHA/MSHA

Regulations: Air Contaminant (29 CFR 1910.1000, Table Z-1, Z-1-A): 5mg/M<sup>3</sup> TWA-8

MSHA: not listed

OSHA Specifically Regulated Substance (29 CFR 1910): not listed

State Regulations: Consult state and local authorities for guidance. Components found in this product may

contain trace amounts of inherent naturally occurring elements (such as, but not limited to arsenic and cadmium) that may be regulated under California Proposition 65 and other

States regulations.

Canada: WHMIS Classification: "D2A" Materials Causing Other Toxic Effects

WHMIS Classification: "E" Corrosive Materials (listed due to corrosive effect on aluminum)

Canada DSL: Listed

# **SECTION 16: OTHER INFORMATION**



NFPA Hazard Class: Health: 2 Flammability: 0 Instability: 0

Physical

HMIS Hazard Class: Health: 2\* Flammability: 0 Hazard: 0 Personal Protection: E

Abbreviations: N/A Not Available or Not Applicable

IARC International Agency for Research on Cancer

IATA International Air Transport Association

ACGIH American Conference of Governmental Industrial Hygienists

TWA Time Weighted Average
PEL Permissible Exposure Limit
TLV Threshold Limit Value

REL Recommended Exposure Limit

dec Decompose